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Convenient preparation of optically active cibenzoline and analogues from 3,3-diaryl-2-propen-1-ols

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ABSTRACT

 $(R)-(+)$ -Cibenzoline (95% ee) was synthesized in two steps from $(+)$ -2,2-diphenylcyclopropylmethanol 3a (98% ee), which was oxidized with IBX in DMSO, followed by treatment with ethylenediamine in the presence of I_2 and K_2CO_3 in tBuOH. Compound $(R)-(+)$ -3a (98% ee) was prepared by cyclopropanation of 3,3-diphenyl-2-propen-1-ol 1 with Et₂Zn and CH₂I₂ in the presence of a catalytic amount of (S) -2-(methanesulfonyl)amino-1-(p-toluenesulfonyl)amino-3-phenylpropane 2, followed by esterification with 3,5-dinitorobenzoyl chloride, recrystallization, and hydrolysis.

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1. Introduction

Commercially available cibenzoline, which is a medicine as a class I antiarrhythmic agent, $\frac{1}{2}$ $\frac{1}{2}$ $\frac{1}{2}$ consists of four rings: a cyclopropane containing a stereogenic carbon, two arene rings, and an imidazoline. A Simmons–Smith reaction is one of the most effective cyclopropane skeleton-forming procedures.[2](#page-5-0) Since the first reports about the enantioselective Simmons–Smith cyclopropanation catalyzed by C_2 -symmetrical disulfonamide-zinc or aluminum complex by Kobayashi, $3e, h-j$ Denmark has optimized conditions for his method^{3d,f,g} while Charette has developed another enantioselective catalytic cyclopropanation using 25 mol % of C_2 -symmetrical titanium complex of (4R,5R)-2,2-diethyl-α,α,α',α'-tetraphenyl-1,3-dioxane-4,5-dimethanol (Ti-TADDOLate).^{3c} Then, we have developed a catalytic enantioselective Simmons–Smith reaction^{3a,b} using the simple disulfonamides derived from α -amino acid. Recently, Katsuki has found a catalytic asymmetric Simmons–Smith reaction using an Al Lewis acid/N Lewis base bifunctional Al Salalen complex.⁴ We have also reported the convenient synthesis of optically active cibenzoline and the analogues.^{[5](#page-6-0)} Herein we report the catalytic enantioselective cyclopropanation of 3,3 diaryl-2-propen-1-ols 1^6 1^6 with Et₂Zn and CH₂I₂ using a catalytic amount of (S)-2-(methanesulfonyl)amino-1-(p-toluenesulfonyl)amino-3-phenylpropane 2 and the synthesis of $(R)-(+)$ -cibenzoline and analogues.

2. Results and discussion

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\begin{picture}(120,140)(-10,0) \put(10,0){\line(1,0){160}} \put(10,0){\line(1,0){160}}
$$

The reaction of 3,3-diphenyl-2-propen-1-ol 1a with $Et₂Zn$ and $CH₂I₂$ in the presence of 10 mol % of 2 afforded the corresponding cyclopropane product $3a^7$ $3a^7$ with 76% ee as indicated in entry 1 of [Table 1](#page-1-0). The cyclopropanation of 3,3-diaryl-2-propen-1-ols substituted on the aromatic ring by electron-donating or electron-withdrawing groups was then examined: the results from the cyclopropanation of various 3,3-diaryl-2-propen-1-ols 1b–1f with Et₂Zn and CH₂I₂ in the presence of 10 mol % of 2 are collected in [Table 1](#page-1-0). We selected methoxy and methyl substituents as representative electron-donating groups (see entries 2 and 3), trifluoromethyl and chloro substituents as electron-withdrawing groups (see entries 4 and 5), and fluorenyl substituent for making the spiro-ring (see entry 6). The cyclopropanation of non-substituted allylic alcohol 1a which can be converted to cibenzoline gave the highest ee among these 3,3-diaryl-2-propen-1-ols 1a–1f. The reactions of the allylic alcohols 1b and 1c (entries 2 and 3, respectively) substituted with an electron-donating group afforded slightly lower enantioselectivities than those of the allylic alcohols 1d and 1e (entries 4 and 5, respectively) substituted with an electron-withdrawing group. The reactions of the allylic alcohol 1f(entry 6) substituted with a tricyclic group also afforded slightly lower enantioselectivity than those of the allylic alcohol 1a (entry 1) non-substituted on the aromatic ring.

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Table 1

^a All reactions were carried out with 1 equiv of 3,3-diaryl-2-propen-1-ol 1, 0.1 equiv of 2, 2 equiv of Et₂Zn, and 3 equiv of CH₂I₂ in anhydrous CH₂Cl₂. b Determined by HPLC analysis using Chiralcel OD.

^c The number indicates the concentration of iPrOH in hexane as an eluent on HPLC analysis for the determination of enantiomeric excess of the product.

trans-3-Phenylbut-2-enol 1g, which possesses different substituents on the 3-position, was converted to the corresponding cyclopropylmethanol $3g^{3c}$ in 85% yield with a low enantiomeric excess, unfortunately.

 $(R)-(+)$ -Cibenzoline **7a**, the absolute configuration and specific rotation of which were already known in the literature,^{1a} was synthesized from (+)-2,2-diphenylcyclopropylmethanol 3a as follows; alcohol 3a was oxidized with 2-iodoxybenzoic acid (IBX) in dimethylsulfoxide (DMSO) at rt for 3 h to afford the corresponding aldehyde 4a, which was treated with NaClO₂, H_2O_2 , and Na H_2PO_4 in MeCN–H₂O at rt for 30 min to give acid $5a$. Acid $5a$ was condensed with ethylenediamine in the presence of benzotriazol-1 yloxytripyrrolidinophosphonium hexafluorophosphate (PyBOP) and Et_3N in CH_2Cl_2 at rt for 5 h to give the corresponding amide 6a in quantitative overall yield from 3a. Then, the amide 6a was

converted under reduced pressure (2 mmHg) at 160° C for 37 h to (R) -(+)-cibenzoline **7a** in 55% yield as shown in Scheme 1. These results suggest that the absolute configurations of the compounds 3a, 4a, 5a, and 6a are R. The fluorenyl alcohol 3f, which is connected at the ortho-position of each benzene ring in 3a, was converted to the corresponding amide 6f in quantitative overall yield in a similar manner to that described for the preparation of 6a as indicated in Scheme 2. Amide 6f was not cyclized under reduced pressure (2 mmHg) at 160 \degree C for 26 h and recovered in 52% yield after silica gel column chromatograpy. It was found that the fluorenyl group of the amide **6f** hinders the stereochemical formation of the 2-imidazoline ring because the two benzene rings of 6f are fixed at the same plane.

Cyclopropylmethanols 3c, 3d, and $3e^8$ $3e^8$ which are oriented with methyl, trifluoromethyl, chloro groups at the para-position on the benzene rings, were converted to the corresponding cibenzoline analogues in a similar manner to that described for the preparation of cibenzoline. All specific rotations of the amides 6a–6f, cibenzoline 7a, and analogues 7c–7e are summarized in [Table 2.](#page-2-0)

The enantiomeric excess of $(R)-(+)$ -cibenzoline **7a** was checked by HPLC analysis with a 80:15:5:0.1 mixture of hexane–ethanol–

Scheme 1. Synthesis of (R) -(+)-cibenzoline 7a from $(+)$ -2,2-diphenylcyclopropylmethanol 3a.

Scheme 2. Trial for the preparation of a cibenzoline analog 7f from $(+)$ -cyclopropylmethanol 3f.

Table 2

Synthesis of $(R)-(+)$ -cibenzoline and analogues **7a** and **7c–7f** from $(+)$ -2,2-diarylcyclopropylmethanols **3a–3f**

^a Overall yields from the cyclopropylmethanols 3a–3f.

^b The enantiomeric excesses were determined by HPLC analysis with a 80:15:5:0.1 mixture of hexane–ethanol–isopropanol–diethylamine as an eluent using Chiralcel OD.1a

isopropanol–diethylamine as an eluent using Chiralcel OD and found to be 15% ee, which is lower than that (76% ee) of the starting alcohol 3a. It is supposed that racemization proceeds under cyclization of the amide 6a at the high reaction temperature. In order to avoid racemization, an alternative route reported by $Togo⁹$ was examined. The aldehydes 4a and 4c-4e were reacted with ethylenediamine in the presence of K_2CO_3 and I_2 in tBuOH to afford the corresponding 2-imidazoline derivatives 7a, 7c–7e in quantitative yields as indicated in Table 3. No racemization was observed in entries 1 and 3–6.

Even when using the fluorenyl type's aldehyde 4f, which hinders steric hinderance, the reaction proceeded to give the cibenzoline analogue 7f in 48% yield. The specific rotation (+186.4) is completely different from that (+75.7) of cibenzoline, hence the biological activity is very interesting.

Next, we tried to improve the enantiomeric purity of the alcohol 3a (76% ee) by recrystallization after the formation of the 3,5-dinitrobenzoyl ester, followed by hydrolysis under basic conditions to afford 98% ee of 3a.^{3a} The alcohol 3a with 98% ee was converted to the aldehyde 4a, which was then cyclized by Togo's method with ethylenediamine, (1R,2R)-1,2-diphenylethylenediamine, or (1R,2R)-cyclohexanediamine to obtain the corresponding 2-imidazolines 7a, 7h, and 7i, respectively, as described in [Scheme 3](#page-3-0) and [Table 4.](#page-3-0)

Table 3

Alternative synthesis of $(+)$ -cibenzoline and analogues 7a and 7c–7f from the aldehydes 4a and 4c–4f

3. Conclusion

In conclusion, (S)-2-(methanesulfonyl)amino-1-(4-toluenesulfonyl)amino-3-phenylpropane 2 efficiently works as a catalyst in the Simmons–Smith cyclopropanation of steric hindered allylic alcohols such as 3,3-diaryl-2-propen-1-ols 1. In our procedure, it is possible to prepare various chiral 3,3-disubstituted 2,3-methano-1-propanols 3 conveniently and to synthesize the corresponding chiral cibenzoline analogues 7 including the imidazoline derivatives. We are still working on the optimization of α -amino acid-derived chiral disulfonamides for the catalytic enantioselective cyclopropanation of 3,3-diphenyl-2-propen-1-ol 1a.

4. Experimental

H

4.1. General

The ¹H NMR spectra were measured with a Bruker UltrashieldTM 400 Plus (400 MHz) spectrometer. The chemical shifts are expressed in ppm downfield from tetramethylsilane (δ = 0.00) as an internal standard. Splitting patterns are indicated as s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; and br, broad signal. The high-resolution Mass spectra (HRMS) of the compounds

H2NCH²CH²NH₂

a The enantiomeric excesses were determined by HPLC analysis with a 80:15:5:0.1 mixture of hexane-ethanol-isopropanol-diethylamine as an eluent using Chiralcel OD.1a

Scheme 3. Synthesis of the enantiopure $(R)-(+)$ -cibenzoline and analogues 7a, 7h, and 7i via recrystallization of the ester 8a.

 a The enantiomeric excess (95% ee) was determined by HPLC analysis with a 80:15:5:0.1 mixture of hexane–ethanol–isopropanol–diethylamine as an eluent using Chiralcel OD.¹⁴

with a high molecular weight were recorded using a Waters LCT Premier (ESI-TOF-MS) spectrometer. Unless otherwise noted, all experiments were carried out under an argon atmosphere. For thin layer chromatographic (TLC) analyses, Merck precoated TLC plates (Silica Gel 60 $F₂₅₄$, Art 5715) were used.

4.2. Typical procedure for Simmons–Smith cyclopropanations

To a colorless clear solution of the sulfonamide 2 (38 mg, 0.5 mmol, 0.1 equiv) and 1 (1 mmol, 1 equiv) in anhydrous <code>CH</code>₂Cl₂ (7.5 mL) was added dropwise at $-40\,^{\circ}$ C a solution of Et₂Zn in hexane (1.05 M, 0.95 mL, 1 mmol, 2 equiv) and $CH₂I₂$ (121 μ L, 1.5 mmol, 3 equiv). After stirring for 3 h at 0 °C, the colorless suspension was quenched at the temperature with Et_3N (0.3 mL), diluted with $Et₂O$ (70 mL), washed with brine (20 mL), and dried over MgSO₄. The crude product was chromatographed on silica gel with a 1:10 mixture of EtOAc and hexane to afford 3.

4.2.1. 2,2-Bis(4-methoxyphenyl)cyclopropylmethanol, 3b

Pale yellow oil; 77% yield; $[\alpha]^{24}$ _D = +74.2 (c 1.04, CHCl₃); 51% ee;
¹H NMR (400 MHz, CDCL) ≥ 14 1.18 (1H, dd, I = 4.8, 8, 8, Hz) 1.26 (1H ¹H NMR (400 MHz, CDCl₃) δ 1.18 (1H, dd, J = 4.8, 8.8 Hz), 1.26 (1H, t, $J = 4.8$ Hz), 1.89 (1H, m), 3.33 (1H, dd, $J = 7.9$, 11.5 Hz), 3.47 (1H, dd, $J = 6.3$, 11.5 Hz), 3.73 (3H, s), 3.76 (3H, s), 6.77 (2H, d, $J = 8.8$ Hz), 6.82 (2H, d, $J = 8.8$ Hz), 7.15 (2H, d, $J = 8.8$ Hz), 7.27 (2H, d, $J = 8.8$ Hz); ¹³C NMR (100 MHz, CDCl₃) δ 17.7, 27.5, 34.2, 55.1, 55.2, 63.8, 113.6, 113.8, 128.8, 130.8, 133.5, 138.9, 157.7, 158.1; HRMS (ESI-TOF): Calcd for $C_{18}H_{20}O_3$ Na (M+Na)⁺: 307.1305, Found: 307.1338.

4.2.2. 2,2-Bis(4-methylphenyl)cyclopropylmethanol, 3c

Colorless oil; 99% yield; $[x]_D^{24} = +101.7$ (c 1.45, CHCl₃); 65% ee;
¹H NMP (400 MHz, CDCL) λ 1.20 (1H dd. I = 4.8, 8.8 Hz), 1.30 ¹H NMR (400 MHz, CDCl₃) δ 1.20 (1H, dd, J = 4.8, 8.8 Hz), 1.30 $(1H, br s)$, 1.33 $(1H, t, J = 4.8 Hz)$, 1.95 $(1H, m)$, 2.28 $(3H, s)$, 2.30 $(3H, s)$, 3.35 (1H, t, $J = 10.9$ Hz), 3.48 (1H, t, $J = 10.9$ Hz), 7.04 (2H, d, $J = 8.2$ Hz), 7.10 (2H, d, $J = 7.9$ Hz), 7.12 (2H, d, $J = 8.2$ Hz), 7.25 (2H, d, J = 7.9 Hz); ¹³C NMR (100 MHz, CDCl₃) δ 17.7, 20.9, 21.0, 27.5, 34.9, 63.9, 127.8, 129.0, 129.3, 129.8, 135.5, 136.2, 138.3, 143.6; HRMS (ESI-TOF): Calcd for C₁₈H₂₀ONa (M+Na)⁺: 275.1406, Found: 275.1389.

4.2.3. 2,2-Bis(4-trifluoromethylphenyl)cyclopropyl-methanol, 3d

Colorless oil; 99% yield; $[\alpha]_D^{24} = +80.0$ (c 1.41, CHCl₃); 73% ee; ¹H NMR (400 MHz, CDCl₃) δ 1.37 (1H, dd, J = 5.2, 9.0 Hz), 1.43 (1H, t, $J = 5.2$ Hz), 1.44 (1H, br s), 2.04 (1H, m), 3.29 (1H, dd, $J = 8.2$, 11.4 Hz), 3.57 (1H, dd, $J = 6.0$, 11.4 Hz), 7.31 (2H, d, $J = 8.1$ Hz), 7.52 (4H, s), 7.59 (2H, d, J = 8.2 Hz); ¹³C NMR (100 MHz, CDCl₃) δ 18.2, 28.2, 35.5, 63.3, 124.08 (q, 1_{C-F} = 272 Hz), 124.11 (q, 1_{C-F} = 272 Hz), 125.5 (q, ${}^{3}J_{C-F}$ = 3.6 Hz), 125.7 (q, ${}^{3}J_{C-F}$ = 3.8 Hz), 128.1, 128.7(q, $^2J_{C-F}$ = 32.4 Hz), 129.4 (q, $^2J_{C-F}$ = 32.6 Hz), 130.8, 144.4, 149.3; HRMS (ESI-TOF): Calcd for $C_{18}H_{14}F_6$ ONa (M+Na)⁺: 383.0841, Found: 383.0871.

4.2.4. 2,2-Bis(4-chlorophenyl)cyclopropylmethanol, 3e⁸

Yellow oil; 88% yield; $[\alpha]_D^{24} = +100.4$ (c 0.78, CHCl₃); 72% ee; ¹H NMR (400 MHz, CDCl₃) δ 1.26 (1H, dd, J = 5.1, 9.0 Hz), 1.28 (1H, br s), 1.32 (1H, t, $J = 5.1$ Hz), 1.95 (1H, m), 3.29 (1H, m), 3.54 (1H, m), 7.13 (2H, d, J = 8.6 Hz), 7.21 (2H, d, J = 8.6 Hz), 7.29 (4H, s); ¹³C NMR (100 MHz, CDCl₃) δ 18.0, 27.8, 34.6, 63.4, 128.5, 128.8, 129.1, 131.5, 132.0, 132.7, 139.2, 144.3.

4.2.5. 2,2-Fluorenylcyclopropylmethanol, 3f

Pale brown oil; 97% yield; $[\alpha]_0^{24} = +12.5$ (c 1.12, CHCl₃); 60% ee;
¹H NMP (400 MHz, CDCL) § 1.32 (1H br s) 1.73 (1H dd J = 5.2 ¹H NMR (400 MHz, CDCl₃) δ 1.32 (1H, br s), 1.73 (1H, dd, J = 5.2, 7.4 Hz), 1.95 (1H, dd, J = 5.2, 9.0 Hz), 2.26 (1H, m), 3.91 (1H, dd, $J = 8.7$, 12.1 Hz), 4.10 (1H, dd, $J = 6.2$, 12.1 Hz), 7.02 (1H, d, $J = 7.0$ Hz), $7.22 - 7.39$ (5H, m), 7.80 (1H, d, $J = 7.0$ Hz), 7.84 (1H, d, $J = 7.5$ Hz); ¹³C NMR (100 MHz, CDCl₃) δ 21.7, 33.5, 61.9, 118.5, 119.7, 120.3, 121.1, 126.1, 126.4, 126.6, 127.1, 139.0, 141.1, 144.1, 148.1; HRMS (ESI-TOF): Calcd for $C_{16}H_{14}$ ONa (M+Na)⁺: 245.0937, Found: 245.0891.

4.2.6. (1S,2S)-(+)-2-Methyl-2-phenylcyclopropyl-methanol, 3g^{3c} Colorless oil; 85% yield; $[\alpha]_D^{27} = +24.1$ (c 1.01, CHCl₃); 38% ee; ¹H NMR (400 MHz, CDCl₃) δ 0.60 (1H, dd, J = 5.0, 5.5 Hz), 1.14 (1H, dd, $J = 5.0, 9.0$ Hz), 1.39-1.45 (1H, m), 1.45 (3H, s), 1.65 (1H, br s), 3.70 $(1H, dd, J = 8.6, 11.5 Hz)$, 3.90 $(1H, dd, J = 6.4, 11.5 Hz)$, 7.15–7.31 (5H, m); ¹³C NMR (100 MHz, CDCl₃) δ 18.7, 20.5, 24.8, 27.8, 63.6, 125.8, 127.2, 128.3, 147.6; HRMS (ESI-TOF): Calcd for $C_{11}H_{14}$ ONa (M+Na)⁺: 185.0937, Found: 185.0917.

4.3. Typical procedure for oxidation of alcohols 3 into the aldehydes 4

To a solution of alcohol 3 in dimethylsulfoxide (DMSO, 2 mL) was added 2-iodoxybenzoic acid (IBX, 4 equiv) at room temperature. After stirring for 3 h at room temperature, EtOAc (10 mL) was added to the reaction mixture. The suspension was filtered. To the filtrate was added half brine, extracted three times with AcOEt. The combined AcOEt layers were washed with brine, dried over anhydrous MgSO4, and evaporated. The crude product was chromatographed on silica gel with a 5:1 mixture of EtOAc and hexane to afford the aldehyde 4.

4.3.1. 2,2-Bis(4-methylphenyl)cyclopropanecarbox-aldehyde, 4c

Colorless oil; 90% yield; $[\alpha]_D^{27} = -77.9$ (c 1.37, CHCl₃); ¹H NMR $(400 \text{ MHz}, \text{ CDCl}_3)$ δ 1.83 (1H, dd, J = 5.2, 8.4 Hz), 2.21 (1H, t, $J = 5.2$ Hz), 2.28 (3H, s), 2.30 (3H, s), 2.49 (1H, ddd, $J = 5.2$, 6.8, 8.4 Hz), $7.05 - 7.13$ (6H, m), 7.27 (2H, m), 8.67 (1H, d, $I = 6.8$ Hz); ¹³C NMR (100 MHz, CDCl₃) δ 20.3, 20.9, 21.1, 36.8, 40.4, 127.2, 129.3, 129.6, 129.8, 136.6, 136.7, 137.1, 141.3, 200.8; HRMS (ESI-TOF): Calcd for $C_{18}H_{18}ONa$ $(M+Na)^{+}$: 273.1250, Found: 273.1258.

4.3.2. 2,2-Bis(4-trifluoromethylphenyl)cyclopropane-carboxaldehyde, 4d

Colorless oil; 89% yield; $[\alpha]_D^{27} = -86.4$ (c 1.43, CHCl₃); ¹H NMR $(400 \text{ MHz}, \text{ CDCl}_3) \delta 1.93 (1\text{H}, \text{ dd}, J = 5.5, 8.5 \text{ Hz}), 2.34 (1\text{H}, \text{t},$ $J = 5.5$ Hz), 2.65–2.70 (1H, m), 7.34 (2H, d, $J = 8.2$ Hz), 7.52 (2H, d, $J = 8.2$ Hz) 7.55 (2H, d, $J = 8.2$ Hz) 7.62 (2H, d, $J = 8.2$ Hz), 8.80 (1H, d, J = 6.2 Hz); ¹³C NMR (100 MHz, CDCl₃) δ 20.3, 36.4, 40.3, 123.80 $(q, 1_{C-F} = 272 Hz)$, 123.84 $(q, 1_{C-F} = 272 Hz)$, 125.9 $(q, 3_{C-F} =$ 3.7 Hz), 126.1 (q, ${}^{3}J_{C-F}$ = 3.7 Hz), 127.8, 129.6 (q, ${}^{2}J_{C-F}$ = 32.8 Hz), 130.2 (q, ${}^{2}J_{C-F}$ = 32.7 Hz), 130.5, 142.4, 146.7, 198.7; HRMS (ESI-TOF): Calcd for $C_{18}H_{12}F_6$ ONa $(M+Na)^+$: 381.0685, Found: 381.0703.

4.3.3. 2,2-Bis(4-chlorophenyl)cyclopropanecarbox-aldehyde, 4e

Yellow solid; 98% yield; Mp 88–89 °C; $[\alpha]_D^{27} = -95.0$ (c 0.65, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 1.84 (1H, dd, J = 5.4, 8.4 Hz), 2.24 (1H, t, $I = 5.4$ Hz), 2.52–2.57 (1H, m), 7.14 (2H, d, $J = 8.6$ Hz), 7.25 (2H, d, $J = 8.6$ Hz), 7.30 (4H, s), 8.73 (1H, d, $J = 6.4$ Hz); ¹³C NMR (100 MHz, CDCl₃) δ 20.3, 36.5, 39.7, 128.7, 128.9, 129.3, 131.3, 133.1, 133.7, 137.4, 141.9, 199.5; HRMS (ESI-TOF): Calcd for $C_{16}H_{12}Cl_2ONa$ (M+Na)⁺: 313.0157, Found: 313.0202.

4.3.4. 2,2-Fluorenylcyclopropanecarboxaldehyde, 4f

Yellow oil; 94% yield; $[\alpha]_{\text{D}}^{27} = -125.2$ (c 0.99, CHCl₃); ¹H NMR $(400 \text{ MHz}, \text{ CDCl}_3)$ δ 2.30 (1H, dd, J = 5.5, 8.3 Hz), 2.66 (1H, dd, $J = 5.5$, 7.4 Hz), 2.90 (1H, ddd, $J = 4.1$, 7.4, 8.3 Hz), 7.09 (1H, d, J = 7.5 Hz), 7.27-7.43 (5H, m), 7.81 (1H, d, J = 6.5 Hz), 7.82 (1H, d, $J = 7.6$ Hz), 9.70 (1H, d, $J = 4.1$ Hz); ¹³C NMR (100 MHz, CDCl₃) δ 20.6, 39.7, 40.4, 118.8, 120.0, 120.2, 122.7, 127.0, 127.3, 127.4, 139.8, 141.0, 141.9, 145.6, 197.0; HRMS (ESI-TOF): Calcd for $C_{16}H_{12}$ ONa (M+Na)⁺: 243.0780, Found: 243.0762.

4.4. Typical procedure for oxidation of the aldehydes 4 into the carboxylic acids 5

To a colorless clear solution of aldehyde 4 and NAH_2PO_4 (0.3 equiv) in MeCN (4 mL) and $H₂O$ (0.4 mL) were added 35% $H₂O₂$ (1.1 equiv) and a solution of NaClO₂ (1.5 equiv) in $H₂O$ (2 mL) at 0 \degree C. After stirring for 30 min at room temperature, saturated $Na₂SO₃$ aq (1 mL) was added to the reaction mixture. Next, 1 M HCl was added, and the reaction mixture was adjusted to pH 3. The reaction mixture was extracted three times with AcOEt. The combined AcOEt layers were washed with brine, dried over anhydrous MgSO4, and evaporated. The obtained the crude carboxylic acid 5 was used in the next step without further purification.

4.5. Typical procedure of the carboxylic acids 5 into the amides 6

To a colorless clear solution of carboxylic acid 5 in anhydrous $CH₂Cl₂$ (2 mL) were added ethylenediamine (20 equiv), triethylamine (3 equiv), and benzotriazol-1-yloxytripyrrolidinophosphonium hexafluorophosphate (PyBOP; 1.2 equiv) at room temperature. After stirring for 5 h at room temperature, water (2 mL) was added to reaction mixture. Then, 1 M NaOH was added, and the reaction mixture was adjusted to pH 10. The reaction mixture was extracted three times with AcOEt. The combined AcOEt layers were washed with brine, dried over anhydrous MgSO4, and evaporated. The crude product was chromatographed on silica gel with a 7:3:0.5 mixture of CHCl₃, MeOH, and H₂O to afford the amine 6.

4.5.1. N-(2-Aminoethyl)-2,2-diphenylcyclopropane-carboxamide, 6a

Pale yellow oil; quantitative yield; $[\alpha]_D^{21} = +102.8$ (c 1.03, MeOH); ¹H NMR (400 MHz, CD₃OD) δ 1.50 (1H, dd, J = 5.0, 8.2 Hz), 2.16 (1H, dd, $J = 5.0$, 5.9 Hz), 2.53 (1H, dd, $J = 5.9$, 8.2 Hz), 2.73 (2H, m), 3.23 (2H, t, $J = 6.2$ Hz), 7.12–7.36 (10H, m); ¹³C NMR (100 MHz, CD₃OD) δ 19.4, 31.0, 38.6, 40.2, 41.1, 127.5, 127.9, 128.7, 129.3, 129.4, 130.9, 142.0, 146.6, 173.6; HRMS (ESI-TOF): Calcd for $C_{18}H_{21}N_2O$ (M+H)⁺: 281.1648, Found: 281.1654.

4.5.2. N-(2-Aminoethyl)-2,2-bis(4-methylphenyl)-cyclopropanecarboxamide, 6c

Colorless oil; quantitative yield; $[\alpha]_D^{21} = +23.9$ (c 1.38, MeOH);
¹H NMP (400 MHz, CD-OD) $\hat{\lambda}$ 1.42 (1H dd. I = 4.8, 8.2 Hz), 2.08 ¹H NMR (400 MHz, CD₃OD) δ 1.42 (1H, dd, J = 4.8, 8.2 Hz), 2.08 $(1H, dd, J = 4.8, 5.9 Hz), 2.23 (3H, s), 2.25 (3H, s), 2.46 (1H, dd,$ $J = 5.9$, 8.2 Hz), 2.74 (2H, m), 3.21 (2H, t, $J = 6.2$ Hz), 7.02 (2H, d, $J = 8.0$ Hz), 7.05 (2H, d, $J = 8.2$ Hz), 7.15 (2H, d, $J = 8.2$ Hz), 7.20 (2H, d, J = 8.0 Hz); ¹³C NMR (100 MHz, CD₃OD) δ 19.3, 20.9, 21.1, 31.2, 39.4, 40.3, 41.4, 128.5, 129.8, 129.9, 130.7, 137.0, 137.4, 139.1, 143.9, 173.2; HRMS (ESI-TOF): Calcd for C₂₀H₂₅N₂O (M+H)⁺: 309.1961, Found: 309.2000.

4.5.3. N-(2-Aminoethyl)-2,2-bis(4-trifluoromethyl-phenyl)cyclopropanecarboxamide, 6d

Colorless oil; 89% yield; $[\alpha]_D^{21} = +19.2$ (c 1.27, MeOH); ¹H NMR (400 MHz, CD₃OD) δ 1.63 (1H, dd, J = 5.5, 8.3 Hz), 2.22 (1H, t, $J = 5.5$ Hz), 2.65 (1H, dd, $J = 5.5$, 8.3 Hz), 2.81(2H, t, $J = 6.3$ Hz), 3.26 (2H, t, J = 6.3 Hz), 7.54 (8H, m); ¹³C NMR (100 MHz, CD₃OD) δ 19.8, 31.1, 39.5, 40.4, 41.2, 125.6 (q, 1 J_{C-F} = 271 Hz), 125.7 (q, 1 _{L- $=$} 271 Hz), 125.3 (q, 3 L_{- $=$} 271 Hz) J_{C-F} = 271 Hz), 126.3 (q, ${}^{3}J_{C-F}$ = 3.7 Hz), 126.5 (q, ${}^{3}J_{C-F}$ = 3.7 Hz), 129.6, 129.9 (q, ${}^{2}J_{C-F}$ = 32.0 Hz), 130.2 (q, ${}^{2}J_{C-F}$ = 32.0 Hz), 131.7, 145.8, 150.1, 172.1; HRMS (ESI-TOF): Calcd for $C_{20}H_{19}F_6N_2O$ (M+H)⁺: 417.1396, Found: 417.1385.

4.5.4. N-(2-Aminoethyl)-2,2-bis(4-chlorophenyl)-cyclopropanecarboxamide, 6e

Pale yellow oil; 55% yield; $[\alpha]_D^{23} = +38.7$ (c 1.39, MeOH); ¹H NMR (400 MHz, CD₃OD) δ 1.49 (1H, dd, J = 5.0, 8.3 Hz), 2.12 (1H, dd, $J = 5.0$, 6.0 Hz), 2.51 (1H, dd, $J = 6.0$, 8.3 Hz), 2.62 (2H, t, $J = 6.4$ Hz), 3.14 (2H, m), 7.28 (8H, m); ¹³C NMR (100 MHz, CD₃OD) d 19.5, 31.3, 38.5, 42.0, 42.8, 129.4, 129.5, 130.4, 132.5, 133.4, 133.7, 140.4, 145.1, 171.8; HRMS (ESI-TOF): Calcd for $C_{18}H_{19}Cl_2N_2O$ (M+H)⁺: 349.0869, Found: 349.0869.

4.5.5. N-(2-Aminoethyl)-2,2-fluorenylcyclopropane-carboxamide, 6f

Colorless oil; quantitative yield; $[\alpha]_D^{23} = +97.4$ (c 1.63, MeOH);
¹H NMR (400 MHz, CD-OD) δ 2.07 (1H, dd. 1 = 5.0, 8.1 Hz), 2.38 ¹H NMR (400 MHz, CD₃OD) δ 2.07 (1H, dd, J = 5.0, 8.1 Hz), 2.38 $(1H, dd, J = 5.0, 8.1 Hz)$, 2.73 $(2H, m)$, 2.79 $(1H, t, J = 8.1 Hz)$, 3.25 $(2H, m)$, 7.14 (1H, d, J = 7.3 Hz), 7.21 (1H, t, J = 7.5 Hz), 7.31 (3H, m), 7.46 (1H, d, $J = 7.8$ Hz), 7.80 (2H, t, $J = 7.5$ Hz); ¹³C NMR $(100 \text{ MHz}, \text{ CD}_3 \text{ OD})$ δ 20.1, 35.4, 37.3, 40.3, 41.2, 120.1, 120.7, 120.9, 123.3, 127.6, 127.8, 127.9, 128.3, 140.8, 142.2, 144.3, 148.2, 171.2; HRMS (ESI-TOF): Calcd for $C_{18}H_{19}N_2O$ (M+H)⁺: 279.1492, Found: 279.1488.

4.6. Typical procedure for cyclization of the amides 6 into (R)- (+)-cibenzoline and analogs 7

Amide 6 was heated at 160 °C under vacuum (2 mmHg) for 37 h. The crude product was chromatographed on silica gel with a 8:2:0.2 mixture of CHCl₃, MeOH, and Et₃N to afford $(R)-(+)$ cibenzoline and analogues.

4.7. Typical procedure for cyclization of the aldehydes 4 into (R)-(+)-cibenzoline and analogues 7

To a solution of the aldehyde 4 (1 equiv) in tBuOH (5 mL) was added a diamine (1.1 equiv). After stirring for 30 min at rt, K_2CO_3 (3 equiv) and I_2 (1.25 equiv) were added to the mixture. The suspension was stirred for 3 h at 70 \degree C under an argon atmosphere, quenched at the temperature with satd $Na₂SO₃$ (4 mL), extracted with AcOEt (3×200 mL), washed with sat. NaHSO₃ (20 mL) and brine (20 mL), and dried over Na₂SO₄. The crude product was pure enough without purification.

4.7.1. $(R)-(+)$ -Cibenzoline 7a¹

Colorless oil; quantitative yield; $[\alpha]_D^{24} = +110.0$ (c 1.12, MeOH); 72% ee; ¹H NMR (400 MHz, CD₃OD) δ 1.58 (1H, dd, J = 5.4, 8.6 Hz), 2.09 (1H, dd, $J = 5.4$, 6.3 Hz), 2.43 (1H, dd, $J = 6.3$, 8.6 Hz), 3.20 (2H, m), 3.37 (2H, m), 7.11-7.38 (10H, m); ¹³C NMR (100 MHz, CD₃OD) d 19.5, 25.2, 30.8, 39.9, 48.9, 127.6, 128.1, 128.8, 129.3, 129.5, 131.0, 141.4, 146.4, 168.9; HRMS (ESI-TOF): Calcd for $C_{18}H_{19}N_2$ (M+H)⁺: 263.1543, Found: 263.1540.

4.7.2. 2-(2,2-Bis(4-methylphenyl)cyclopropyl)-imidazoline 7c

Colorless oil; quantitative yield; $[\alpha]_D^{23} = +66.0$ (c 0.90, MeOH); 65% ee; ¹H NMR (400 MHz, CD₃OD) δ 1.61 (1H, dd, J = 5.9, 8.6 Hz), 2.08 (1H, t, J = 5.9 Hz), 2.25 (3H, s), 2.28 (3H, s), 2.46 (1H, dd, J = 5.9, 8.6 Hz), 3.29 (2H, m), 3.46 (2H, m), 7.04 (2H, d, $J = 8.1$ Hz), 7.09 (2H, d, $J = 8.0$ Hz), 7.15 (2H, d, $J = 8.1$ Hz), 7.23 (2H, d, J = 8.0 Hz); ¹³C NMR (100 MHz, CD₃OD) δ 19.7, 21.0, 21.1, 25.1, 39.6, 48.5, 128.5, 129.9, 130.0, 130.8, 137.3, 137.9, 138.3, 143.4, 169.3; HRMS (ESI-TOF): Calcd for $C_{20}H_{23}N_2$ (M+H)⁺: 291.1856, Found: 291.1863.

4.7.3. 2-(2,2-Bis(4-trifluoromethylphenyl)cyclopropyl)-imidazoline 7d

Colorless oil; quantitative yield; $[\alpha]_D^{23} = +68.6$ (c 1.18, MeOH); 74% ee; ¹H NMR (400 MHz, CD₃OD) δ 1.82 (1H, dd, J = 6.2, 8.7 Hz), 2.29 (1H, t, $J = 6.2$ Hz), 2.69 (1H, dd, $J = 6.2$, 8.7 Hz), 3.35 (2H, m), 3.51 (2H, m), 7.57 (8H, m); ¹³C NMR (100 MHz, CD₃OD) δ 19.7, 25.1, 39.6, 125.6 (q, 1 J_{C-F} = 271 Hz), 126.5 (q, 3 J_{C-F} = 3.7 Hz), 126.6 $(q, {}^{3}J_{C-F} = 3.7 \text{ Hz})$, 129.5, 130.2 $(q, {}^{2}J_{C-F} = 32.7 \text{ Hz})$, 130.7 $(q, {}^{2}J_{C-F} =$ 32.7 Hz), 131.9, 144.7, 149.4, 168.2; HRMS (ESI-TOF): Calcd for $C_{20}H_{17}F_6N_2$ (M+H)⁺: 399.1290, Found: 399.1302.

4.7.4. 2-(2,2-Bis(4-chlorophenyl)cyclopropyl)imidazoline 7e

Colorless oil; quantitative yield; $[\alpha]_D^{24} = +86.0$ (c 0.97, MeOH); 71% ee; ¹H NMR (400 MHz, CD₃OD) δ 1.88 (1H, dd, J = 6.3, 8.6 Hz), 2.28 (1H, t, $J = 6.3$ Hz), 2.76 (1H, dd, $J = 6.3$, 8.6 Hz), 3.53 $(2H, m)$, 3.71 $(2H, m)$, 7.30 $(2H, d, J = 8.9 Hz)$, 7.34 $(2H, d, J)$ $J = 8.9$ Hz), 7.37 (4H, s); ¹³C NMR (100 MHz, CD₃OD) δ 20.0, 23.7, 40.7, 46.2, 129.9, 130.1, 130.4, 132.4, 134.2, 135.0, 138.5, 143.3, 170.0; HRMS (ESI-TOF): Calcd for $C_{18}H_{17}Cl_2N_2$ (M+H)⁺: 331.0763, Found: 331.0769.

4.7.5. 2-(2,2-Fluorenylcyclopropyl)imidazoline 7f

Colorless solid; 48% yield; Mp 151-166 °C; $[\alpha]_D^{27} = +186.4$ (c 0.34, CHCl₃); 58% ee; ¹H NMR (400 MHz, CD₃OD) δ 2.11 (1H, dd, $J = 5.1$, 8.6 Hz), 2.50 (1H, dd, $J = 5.1$ $J = 7.4$ Hz), 2.57 (1H, t, $J = 8.0$ Hz), 3.46–3.57 (2H, m), 3.59–3.74 (2H, m), 4.76 (1H, br s), 7.07 (2H, d, J = 7.5 Hz), 7.23-7.46 (5H, m), 7.82 (2H, d, J = 7.5 Hz); ¹³C NMR (100 MHz, CD₃OD) δ 20.3, 27.9, 29.7, 35.1, 49.1, 118.6, 119.9, 120.0, 121.3, 126.8, 126.9, 127.1, 139.8, 140.6, 143.3, 146.7, 164.2; HRMS (ESI-TOF): Calcd for $C_{18}H_{17}N_2$ (M+H)⁺: 261.1386, Found: 261.1376.

4.7.6. (4R,5R,1'R)-2-(2,2-Diphenylcyclopropyl)-4,5-diphenyl-2 imidazoline 7h

Yellow oil; 99% yield; $[\alpha]_D^{27} = -152.5$ (c 1.00, MeOH); ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 1.63 (1H, dd, J = 5.2, 8.7 Hz), 2.34 (1H, dd, $J = 5.2, 6.2$ Hz), 2.69 (1H, dd, $J = 6.2, 8.7$ Hz), 4.30 (1H, br s), 4.68 (1H, br s), 6.60 (1H, br s), 7.17-7.52 (20H, m); ¹³C NMR $(100 \text{ MHz}, \text{ CDCl}_3)$ δ 18.8, 25.0, 38.3, 77.2, 126.5, 126.6, 127.0, 127.8, 128.4, 128.5, 128.8, 130.2, 140.3, 143.1, 145.3, 163.7; HRMS (ESI-TOF): Calcd for $C_{30}H_{27}N_2$ (M+H)⁺: 415.2169, Found: 415.2203.

4.7.7. (4R,5R,1'R)-2-(2,2-Diphenylcyclopropyl)-4,5-cyclohexanimidazoline 7i

Yellow solid; 65% yield; Mp 155–165 °C; $[\alpha]_D^{27} = -177.1$ (c 0.71, MeOH); ¹H NMR (400 MHz, CDCl₃) δ 1.07-1.35 (4H, m), 1.53 (1H, dd, $J = 5.3$, 8.8 Hz), 1.58–1.73 (2H, m), 1.94–2.07 (2H, m), 2.19 $(1H, t, J = 5.7 Hz)$, 2.53 $(1H, dd, J = 6.3, 8.8 Hz)$, 2.49–2.63 $(4H, m)$, 3.12 (1H, br s), 7.16–7.30 (10H, m); ¹³C NMR (100 MHz, CDCl₃) δ 18.6, 24.8, 25.8, 30.5, 37.5, 69.1, 126.5, 126.7, 128.07, 128.11, 128.5, 129.5, 140.1, 145.0, 166.2; HRMS (ESI-TOF): Calcd for $C_{22}H_{25}N_2$ (M+H)⁺: 317.2012, Found: 317.1969.

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